



## Variations of chemical composition of NR-PM<sub>1</sub> under the influence of sea land breeze in a coastal city of Southeast China

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### ABSTRACT

Sea land breeze (SLB) is a common local mesoscale circulation in coastal areas, which varies local weather conditions and further affects the diffusion and transport of air pollutants. This study investigated the variation of meteorological conditions, air pollutants concentrations, and aerosol chemical composition under the influence of SLB circulation in a coastal city of Southeast China during Nov. 1 – Dec. 31, 2020, when SLB circulation frequently occurred. The day-night difference in meteorological parameters was amplified on SLB days with a marked daytime peak of *T* and UV while the nighttime pronounced minimum boundary layer height compared to non-SLB days. The average mass concentrations of NR-PM<sub>1</sub> (non-refractory submicron particles) measured by a Q-ACSM were  $9.6 \pm 6.5 \mu\text{g m}^{-3}$  on SLB days and  $11.6 \pm 7.3 \mu\text{g m}^{-3}$  on non-SLB days, but the NR-PM<sub>1</sub>, as well as gaseous pollutants like NO<sub>2</sub> on SLB days, showed a marked peak in the late evening. The NR-PM<sub>1</sub> on SLB days was characterized by a high fraction of organic aerosol (OA). Moreover, the NR-PM<sub>1</sub> showed a significant increase of more-oxidized OOA (MO-OOA) during the sea breeze period, which was associated with enhanced photochemical reactions due to strong UV and elevated O<sub>x</sub>. Comparison between a SLB case and a local pollution case further highlighted the elevated contribution of OOA to NR-PM<sub>1</sub> and the cyclic amplification of air pollutants in SLB conditions. Our study enhances the understanding of the influence of SLB on air pollutants and aerosol chemical composition and provide a plausible explanation for the atmospheric pollution processes in coastal cities.

### 1. Introduction

Aerosols have both direct and indirect effects on global climate (Bellouin et al., 2020; Cohen et al., 2017; Paasonen et al., 2013) and have drawn wide attention in the past few decades owing to its effects on air quality (Lelieveld et al., 2015). Furthermore, aerosols, especially submicron aerosols, are harmful to human health (Shiraiwa et al., 2017). The loading of aerosol present in the atmosphere is directly related to the simultaneous primary and secondary aerosol precursor emissions from multiple sources, secondary aerosol formation rates, removal processes (e.g., sedimentation and precipitation scavenging), and the synergies of local meteorological conditions and regional transport (An et al., 2019; Hu et al., 2017; Sun et al., 2016).

Sea land breeze (SLB) is a local mesoscale topographical circulation unique to the atmospheric boundary layer in coastal areas. The study of SLB first began in the 1920s when Jeffreys put forward the theory of the generation of SLB (Jeffreys, 1922). SLB not only affects local weather variations but sometimes also triggers strong convective weather such as thunderstorms, strong winds, and short-term heavy precipitation along the sea breeze front, which affects the diffusion and transport of air pollutants in the region (Ding et al., 2004; Liu and Chan, 2002; Miao and Yang, 2020). SLB circulation is a wind system that varies within a natural day. The temperature of land is higher than that from the sea during the day, and it is reverse at night, resulting in the circulation characteristics of sea breeze (SB) during the day and land breeze (LB) at night. The LB typically begins at midnight and continues until after sunrise,

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resulting in the urban air masses being transported to the marine in darkness for several hours (Cass and Shair, 1984).

The temperature inversion formed by the encounter of coastal land-sea air currents is an important unfavorable meteorological condition leading to particulate matter pollution, which inhibits vertical movement and transport of air due to the lack of thermal convection and sustained stable stratification (Yang et al., 2022). What's more, the pollutants accumulation of the same air mass repeating across the observation site due to the daily LB/SB reversals probably results in high aerosol concentrations and enhanced contributions of aging components (Cass and Shair, 1984; Lo et al., 2006). Nocturnal chemical transformations are probably important in determining the overall impact of LB on air quality, especially considering that the products of nocturnal reactions are blown back inland by SB during the following day (Wagner et al., 2012). Augustin et al. (2020) pointed out that the SLB circulations provide favorable conditions for secondary aerosol (SA) formation, such as temperature/humidity fluctuations, boundary layer circulation, and relatively high concentrations of various precursors. Previous studies have often applied a theoretical approach using analytical and numerical modeling to study the physical processes responsible for the SLB development and the spatiotemporal distribution of pollutants (Liu et al., 2022; Miller et al., 2003; Papanastasiou and Melas, 2009; Tsai et al., 2011). However, field observations about the variation of aerosol chemical composition under the influence of SLB are still limited (Di Bernardino et al., 2021).

Here, the study on the influence of SLB circulations on air pollutants, as well as aerosols and its chemical composition was conducted in Xiamen, a coastal city of Southeast China, where SLB commonly occurs. The average number (percentage) of SLB days was 60 (14.5–20.8%) per year during the period 2017–2020, which frequently presented in autumn and winter season (Fig. S1). Meteorological parameters, conventional air pollutants, and the chemical composition of NR-PM<sub>1</sub> (non-refractory submicron particles) were simultaneously observed during the period with frequent SLB occurrence and high aerosol concentrations. The main objectives of this study were to (1) reveal the variations of meteorology and conventional pollutants from non-SLB to SLB days, (2) characterize the NR-PM<sub>1</sub> and its chemical composition on SLB days and non-SLB days, and (3) investigate the evolution of air pollutants and aerosol chemical composition during the SLB process.

## 2. Experimental methods

### 2.1. Overview of field observations

Xiamen is located on the west coast of the Taiwan Strait (Fig. 1) and is a subtropical city with a warm climate and high relative humidity throughout the year. The observation period in this study was from Nov. 1 to Dec. 31, 2020, when the SLB phenomenon frequently occurred. Two days (Nov. 5–6) were excluded for instrument malfunction.

The field measurements were conducted at the Institute of Urban Environment (IUE), Chinese Academy of Sciences (118°03'E, 24°36'N) in Jimei District of Xiamen city. IUE is a suburban site close to Jimei Avenue and Haixiang Avenue with high traffic flow (~100 m away). Industrial point sources mainly distributed to the northeast and the southwest of the study site (> 5 km away). The IUE site was located approximately 17 km off the southeast coast of China. The instruments were deployed on the top of the 80 m building at the IUE site.

### 2.2. Observation instruments

The chemical composition of NR-PM<sub>1</sub>, including organics aerosol (OA), nitrate (NO<sub>3</sub>), sulfate (SO<sub>4</sub>), ammonium (NH<sub>4</sub>), and chloride (Cl), was measured by an ACSM. The detail of ACSM instrument operation and calibration can be found in previous studies (Ng et al., 2011; Sun et al., 2012; Chen et al., 2022). The ionization efficiency (IE) was  $4.27 \times 10^{-11}$  and relative ionization efficiency (RIE) for ammonium and sulfate

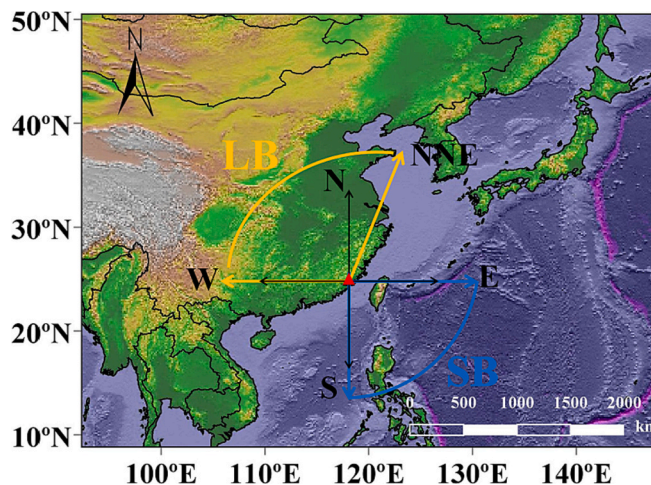


Fig. 1. Location of Xiamen (red triangle) in China. The sea breeze (SB) is blowing from the east-south (E-S), and the land breeze (LB) is blowing from the west-north-northeast (W-N-NNE). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

was 5.49 and 0.53, respectively. The default RIE values were used for nitrate (1.1), OA (1.4), and chloride (1.3) (Canagaratna et al., 2007; Ng et al., 2011).

Trace gases O<sub>3</sub>, NO<sub>x</sub>, CO, and SO<sub>2</sub> were simultaneously measured by Thermo Fisher 49i, 42i, 48i, and 43i, respectively (Thermo Fisher Scientific, Waltham, MA, USA). Those instruments were regularly calibrated and maintained to keep well operational status and data quality. Meteorological parameters, i.e., wind speed (WS), wind direction (WD), temperature (T), relative humidity (RH), and pressure (P), were recorded from an automatic weather station of Xiamen, about 0.9 km far away from IUE. Ultraviolet radiation (UV) was determined by a UV radiometer (KIPP & ZONEN, SUV5 Smart UV Radiometer). Boundary layer height (BLH) was obtained from the European Centre for Medium-Range Weather Forecasts (ECMWF) reanalysis (<https://www.ecmwf.int>).

### 2.3. Identification of OA factors

The data of ACSM was analyzed with the standard Wave Metrics Igor Pro based data analysis software (version 6.37). The data analysis protocols were referred to the previous studies (Sun et al., 2012; Wang et al., 2017). The collection efficiency (CE) values were calculated using algorithms described by Middlebrook et al. (2012). The yielded composition-dependent CEs ( $0.51 \pm 0.07$ ) were comparable to the empirical CE of 0.5. In this study, we adopted the CE empirical value of 0.5, which has been widely used in field observations (Sun et al., 2015; Xu et al., 2017; Zhao et al., 2017; Zhao et al., 2020).

Five OA factors including HOA (hydrocarbon-like OA), BBOA (biomass burning OA), CCOA (coal combustion OA), LO-OOA (less-oxidized oxygenated OA), and MO-OOA (more-oxidized oxygenated OA), were identified by SoFi (version 6.G) along with the multi-linear engine (ME-2) algorithm (Canonaco et al., 2013). In this study, we used the spectra profile of HOA and BBOA derived from standard spectra as constraints to analyze the source of organic matrices during the whole observation period (Ng et al., 2011), and other factors were not constrained. The constraint value (a-value) was selected from 0 to 1 with 0.1 as an interval. Note that the average spectra with constraint values of 0, 0.1, and 0.2 are used for analysis in this study.

The mass spectral and identification of OA factors are presented in SI (Fig. S2). Specifically, CCOA was mixed with COA (cooking OA) due to the strong correlation between CCOA and COA fragments ( $r = 0.96$  for  $m/z$  55 and  $r = 0.68$  for  $m/z$  98). The distinction between LO-OOA and MO-OOA is their degree of oxidation (Xu et al., 2017), and the fraction

of  $m/z$  43 was larger for LO-OOA than MO-OOA. Consistent with the lower degree of oxidation for LO-OOA, its correlations with hydrocarbon-like ions (e.g.,  $C_nH_{2n+1}^+$  and  $C_nH_{2n-1}^+$ ) were stronger than those for MO-OOA. In addition, LO-OOA correlated well with both cooking and combustion fragments, indicating that LO-OOA contained high levels of freshly emitted species (Hu et al., 2016).

### 3. Results and discussion

#### 3.1. Identification of sea land breeze (SLB)

Sea land breeze is a mesoscale local circulation in the lower atmosphere caused by the difference in the thermal properties of land and sea (Masselink and Pattiaratchi, 1998). The identification and definition of SLB are not consistent at different sites, depending on the local weather conditions and the background synoptic winds (Gahmberg et al., 2010). Differences in latitude, coastline shape, and coastal topography lead to SLB with commonalities and local characteristics in different regions (Borne et al., 1998). In this study, the coastline is characterized by a northeast-southwest trend. The sea breeze (SB) is blowing from the east-south (E-S), and the land breeze (LB) is blowing from the west-north-northeast (W-N-NNE). Still winds (wind speed  $<0.2 \text{ m s}^{-1}$ ) are neither sea breezes nor land breezes.

According to the observed wind direction, we firstly divided a natural day into four periods. Specifically, 01:00–08:00 LT is the LB period, 13:00–20:00 LT is the SB period, and 9:00–12:00 LT and 21:00–24:00 LT are the LB and SB transition periods. As shown in the workflow (Fig. S3), the days met with the following conditions were counted as SLB days: (1) The 24-h average ground wind speed is below  $10 \text{ m s}^{-1}$ , (2) During the LB period, the occurrence hour of land breeze is  $\geq 4 \text{ h}$ , and the occurrence hour of sea breeze is  $\leq 2 \text{ h}$ , and (3) During the SB period, the occurrence hour of sea breeze is  $\geq 4 \text{ h}$ , and the occurrence hour of land breeze is  $\leq 2 \text{ h}$ .

Based on the above method, a total of 11 SLB days (Fig. 2, pink area

marked) were identified from November to December 2020. Other days were classified as non-SLB days. Meteorological parameters such as wind speed, wind direction, and boundary layer height that differed from non-SLB days to SLB days probably affect variations in gaseous pollutants and aerosol chemical composition, which would be discussed in detail later.

#### 3.2. Variations of meteorology and conventional pollutants

Relevant studies have shown that the sea-land temperature difference is the major factor driving SLB phenomenon. The larger the temperature difference between land and sea, the stronger the development trend of SLB (Furberg et al., 2002). Consistently, the diurnal variations of temperature showed a larger daytime and nighttime difference on SLB days than on non-SLB days (Fig. 3e). There was little difference in average RH between SLB and non-SLB days (Table S1). However, when wind was transitioning to SB after 13:00, the RH on SLB days remarkably increased compared to non-SLB days, indicating that the SB carried sufficient water vapor. High RH at night during SLB days had a potential effect on the aqueous-phase secondary aerosol formation (Wang et al., 2017; Xu et al., 2017).

The parameters related to photochemical reaction, such as UV and  $O_x$  ( $O_3 + NO_2$ ) were also higher on SLB days than non-SLB days, especially for the daytime (Fig. 3d). The peaked  $T$  and UV during the daytime would be conducive to promote the process of chemical reactions (Chen et al., 2021; Zhou et al., 2020). Although the average mixing ratio of  $O_3$  was slightly lower on SLB days, the diurnal variations showed that the  $O_3$  mixing ratio during the daytime was higher on SLB days than non-SLB days (Fig. 3f). While, the major source of  $O_3$  is photochemical reaction formation and transport (Huang et al., 2006; Zhao et al., 2016). The marked  $O_3$  peak at midday during SLB days was likely resulted from the enhanced photochemical formations due to the higher UV. When the intensity of photochemical reactions weakened, the  $O_3$  mixing ratio decreased more slowly on non-SLB days compared to SLB days, which

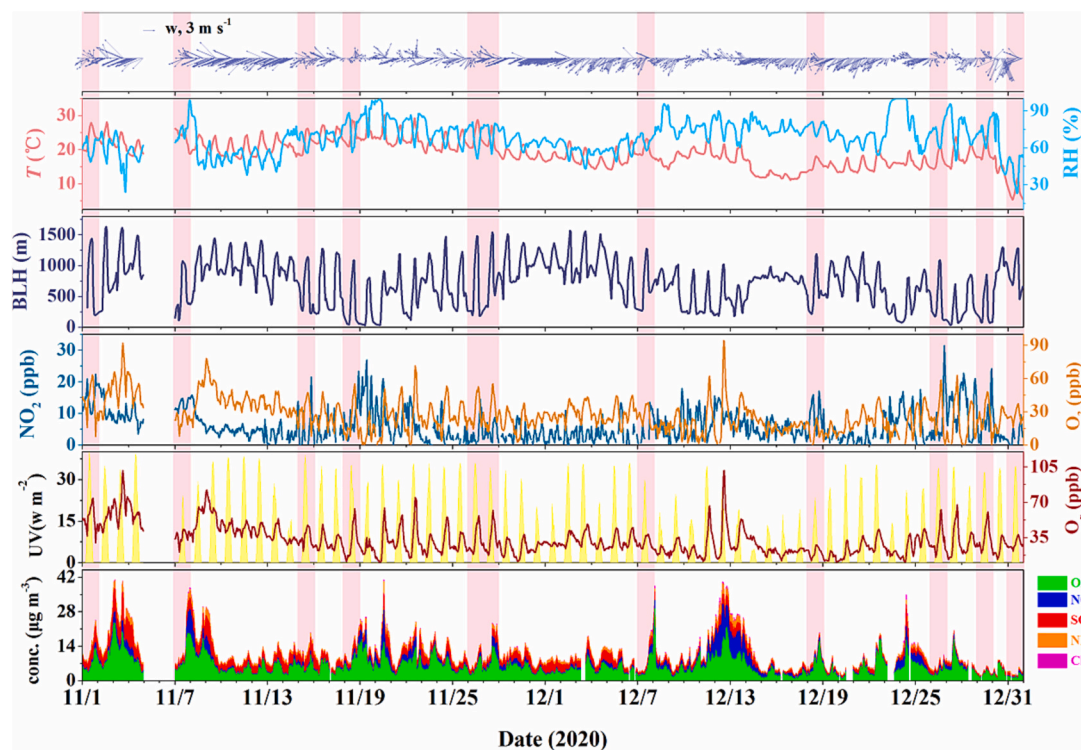


Fig. 2. Temporal variations of meteorological parameters, gaseous pollutants, and chemical composition (OA: organic aerosol,  $NO_3$ : nitrate,  $SO_4$ : sulfate,  $NH_4$ : ammonium, and Cl: chloride) in NR- $PM_{10}$  during the observation period. The pink area is the day of the sea land breeze (SLB) days. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

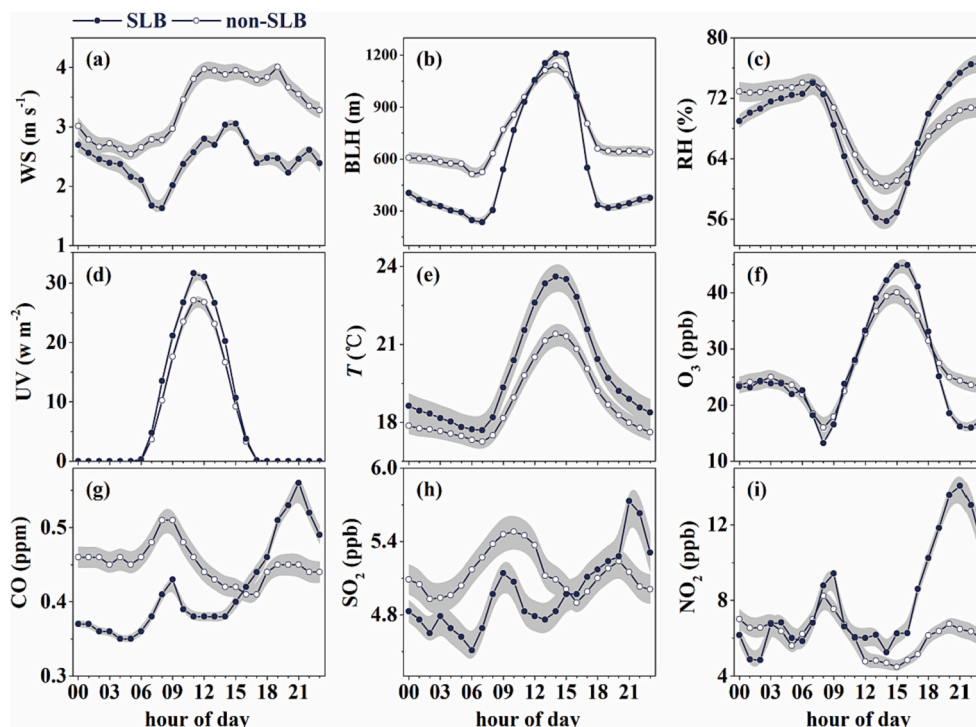


Fig. 3. Average diurnal variations of meteorological parameters ((a) WS, (b) BLH, (c) RH, (d) UV, (e) T) and mixing ratios of gaseous pollutants ((f) O<sub>3</sub>, (g) CO, (h) SO<sub>2</sub>, (i) NO<sub>2</sub>) during SLB (solid point) and non-SLB (hollow point) days. Note: The error bar value is one-tenth of the original value.

could be attributed to the effect of a stronger transport on non-SLB days (Fig. S4). O<sub>x</sub> was proven to be a robust indicator of atmospheric oxidation capacity (Herndon et al., 2008; Xu et al., 2017). We expected that combined the more favorable meteorology (high UV) and enhanced oxidation capacity (elevated O<sub>x</sub>) on SLB days would have significant effect on the photochemical formations of secondary components, like NO<sub>3</sub>, SO<sub>4</sub>, and OOA (Sun et al., 2016; Yang et al., 2020; Xu et al., 2017).

Among the observed meteorological parameters, the BLH and WS showed the most remarkable differences between SLB and non-SLB days (Table S1). The average BLH and WS were  $560 \pm 394$  m and  $2.44 \pm 1.12$  m s<sup>-1</sup> on SLB days relative to  $743 \pm 353$  m and  $3.33 \pm 1.42$  m s<sup>-1</sup> on non-SLB days, respectively. The lower BLH and WS during SLB days were not conducive to the diffusion of pollutants. Accordingly, the mixing ratio of NO<sub>2</sub>, which is mainly emitted from local traffic and industries, increased during SLB periods. The gaseous pollutants like CO, SO<sub>2</sub>, and NO<sub>2</sub> (Fig. 3g, h, i) during non-SLB days showed a bimodal peak in the morning and evening rush hours. In contrast, the concentration of those pollutants during SLB days increased rapidly from the late afternoon with the remarkable evening peak. The BLH decreased rapidly in

the afternoon during SLB days compared to non-SLB days (Fig. 3b). Thus, the late evening peak of pollutants on SLB days was likely a combined effect of pollutant emissions during the evening rush hours and the rapid decline of BLH. Another possible reason was that the contaminants being carried to the ocean by the LB would be brought back to land by the developed SB, which could contribute to the late evening peaks of gaseous pollutants during SLB days.

### 3.3. Variations of NR-PM<sub>1</sub> and chemical composition

#### 3.3.1. Comparison of NR-PM<sub>1</sub> on SLB and non-SLB days

The average mass concentrations of NR-PM<sub>1</sub> on SLB and non-SLB days were  $9.6 \pm 6.5$  μg m<sup>-3</sup> and  $11.6 \pm 7.3$  μg m<sup>-3</sup>, respectively. There was 16.5% lower on SLB than on non-SLB days (Table S1). As shown in Fig. 4a, the diurnal variation trend of NR-PM<sub>1</sub> on non-SLB days was flat. In contrast, the concentration of NR-PM<sub>1</sub> during SLB days showed a low initial level in the early morning, increased continuously during the daytime and peaked in the late evening. On the one hand, when the LB was transformed into SB, the SB brought back the polluted

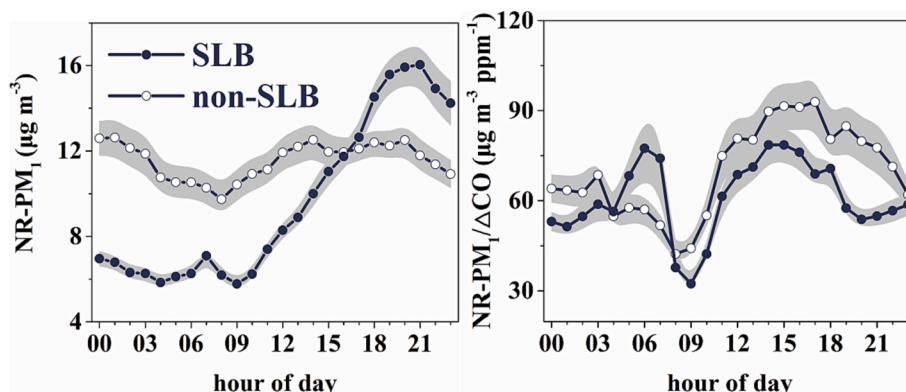


Fig. 4. Diurnal variations of NR-PM<sub>1</sub> mass concentrations (a) and NR-PM<sub>1</sub>/ΔCO ratios (b) during SLB and non-SLB days.

air mass that blew to the ocean during the LB. Alternating LB and SB circulation allowed the air mass carrying pollutants to pass through the observation site multiple times. On the other hand, the shallow boundary layer in the evening (after 18:00, Fig. 3b) was expected to have a significant impact on pollutant accumulation. During the SB period, the stable over-water air mass advection onto land, the height of the boundary layer in coastal areas decreases significantly due to the formation of the thermal internal boundary layer (TIBL; Berman et al., 1999). TIBL made the atmospheric boundary layer more stably stratified and unfavorable diffusion of pollutants with strong sinking flowing (Yang et al., 2022). At this time, the downdraft air carries pollutants to the surface and inhibits the vertical diffusion of pollutants that continue to be emitted from the terrestrial surfaces. The redistribution of pollutants thus led to higher ground concentrations. In addition, studies have shown that low-level temperature inversion is prone to occur during the SB period, which inhibits the diffusion of pollutants (Yang et al., 2022). A previous study in Hong Kong, China, reported that the accumulation of local and regional pollutants was enhanced by slowing surface winds and a well-defined SLB establishment (Lee et al., 2013).

In this study,  $\Delta\text{CO}$  was used to normalize the mass concentration of NR-PM<sub>1</sub> to deduct the effect of physical conditions such as the BLH and the partial effect of primary emissions. Different to absolute NR-PM<sub>1</sub> concentrations, the diurnal variation of NR-PM<sub>1</sub>/ $\Delta\text{CO}$  ratios was more similar between SLB and non-SLB (Fig. 4b). NR-PM<sub>1</sub>/ $\Delta\text{CO}$  ratios increased in the daytime likely due to the photochemical formation of aerosols. However, in the evening, the NR-PM<sub>1</sub>/ $\Delta\text{CO}$  ratios on non-SLB days showed a continuous downward trend, while the ratio on SLB days decreased to evening rush hours and then increased to a peak before sunrise. The increase in NR-PM<sub>1</sub>/ $\Delta\text{CO}$  ratios at night on SLB days was probably ascribed to the influence of nocturnal chemistry. Distinctly, the outstanding peak of NR-PM<sub>1</sub> concentrations in the evening (19:00–21:00) on SLB days was in contrast to the relatively low NR-PM<sub>1</sub>/ $\Delta\text{CO}$  ratios. The comparison indicates that the late evening peak of NR-PM<sub>1</sub> on SLB days was largely attributed to the collapse of the atmospheric boundary layer that favors the accumulation of pollutants.

### 3.3.2. Comparison of chemical composition on SLB and non-SLB days

In terms of the chemical composition of NR-PM<sub>1</sub>, the concentrations of major chemical composition were basically lower on SLB days than non-SLB days, with the smallest change for OA (−7.2%), largest change for NO<sub>3</sub> (−36.9%), and moderate change for other composition (−23.8% – −20.7%) (Table S1). The chemical composition of NR-PM<sub>1</sub> was slightly different from SLB to non-SLB. The former had lower SIA (sulfate, nitrate, and ammonium) and higher OA fraction than the latter (Fig. S5). As shown in Fig. S6, the reduction of SO<sub>4</sub> fraction on SLB days evolved from late afternoon to evening. Sulfate is formed over a regional scale and is generally considered to be transportable over long distances (Shen et al., 2012; Zhou et al., 2020). SLB is a local mesoscale topographical circulation, thereby, the impact of transport on observation sites is relatively low on SLB days, which could explain the reduction of sulfate fraction on SLB days. The relatively low fraction of NO<sub>3</sub> on SLB days could be attributed to the rapid decomposition of NO<sub>3</sub> with higher temperatures and radiation (Seinfeld and Pandis, 2016; Sun et al., 2012).

The fractions of OA factors all increased to varying degrees on SLB days. The increased POA fraction (HOA, BBOA, and CCOA) demonstrated that the SLB phenomenon was in favor of the local accumulation of pollutants. The increase in OOA (LO-OOA and MO-OOA) was likely due to enhanced formation processes characterized by strong UV and atmospheric oxidation capacity during SLB days as mentioned above. A previous study conducted on Pearl River Delta (PRD) has observed an increase in O/C ratio on a SLB circulation, indicating a considerable rise in the oxidation degree of OA in this period (Lee et al., 2013). Additionally, the chemical composition of NR-PM<sub>1</sub> during the time period 19:00–21:00 of SLB days was characterized by a significant increase of HOA and NO<sub>3</sub> fractions (Fig. S6). HOA and NO<sub>3</sub> are the primary and

converted pollutants mainly from traffic emissions. Simultaneously, the HOA/ $\Delta\text{CO}$  ratio exhibited a peak during evening rush hours (Fig. S7e) and the nitrate/ $\Delta\text{CO}$  ratio continued to rise after afternoon peak, especially after the evening rush hours (Fig. S7a). The continuously increase in the nitrate/ $\Delta\text{CO}$  ratio before midnight could be contributed by the enhanced nocturnal N<sub>2</sub>O<sub>5</sub> heterogeneous uptake reactions (Sun et al., 2018; Wang et al., 2017). Thus, the result may suggest that the primary pollutants and precursors emitted from traffic sources largely contributed to the marked evening peak of NR-PM<sub>1</sub> concentrations.

### 3.3.3. Variation of chemical composition on SLB days

During SLB days, NR-PM<sub>1</sub> concentrations were 2 times higher within SB periods (12.5  $\mu\text{g m}^{-3}$ ) than LB periods (6.4  $\mu\text{g m}^{-3}$ ). The chemical composition of NR-PM<sub>1</sub> varied significantly from LB to SB, especially for sulfate and MO-OOA (Fig. 5). The fraction of sulfate evidently decreased from 26.4% in LB periods to 19.8% in SB periods. Many studies revealed that sulfate was effectively formed through the cloudy and aqueous chemistry (Harris et al., 2013; Seinfeld and Pandis, 2016), and the elevated sulfate during the day was probably caused by strong transport and advection in the afternoon (Cao et al., 2017; Massimi et al., 2022; Salcedo et al., 2006). In this study, the high sulfate fraction in LB periods was contributed by enhanced aqueous-phase reactions favored by high RH at night (1:00–8:00, Fig. 3c). On the other hand, SLB is a mesoscale local circulation, thus on SLB days, the transport unlikely contributed much to sulfate during the daytime (i.e. SB periods) as mentioned above.

As shown in Fig. S8, the concentration of MO-OOA increased the most among all the chemical composition during the SB periods compared to LB periods. The fraction of MO-OOA increased from 15.5% in LB periods to 21.1% in SB periods, while the fraction of LO-OOA remained comparable between the two periods (Fig. 5). The previous studies had shown that the dominant formation pathway of LO-OOA was photochemical reaction (Hu et al., 2016; Xu et al., 2017), but the high O<sub>x</sub> condition was favorable for the further conversion of LO-OOA to MO-OOA (Sun et al., 2011; Zhan et al., 2021). The later and broader peak of the MO-OOA/ $\Delta\text{CO}$  than that of the LO-OOA/ $\Delta\text{CO}$  also suggests the conversion of LO-OOA to MO-OOA on SLB days with high O<sub>x</sub> (Fig. S7h, i). The remarkable increase of MO-OOA in SB periods was likely explained by efficient oxidation of POA and LO-OOA under SLB circulatory conditions. This result was expected because of the strong UV and low RH environment (Fig. 3c, d) during SLB-SB period favorable for photochemical processes and aerosol aging due to air mass retention.

### 3.4. Evolution of a SLB case

To further investigate the influence of SLB circulation on aerosols, we compared a SLB case and a non-SLB case in detail (Fig. 6). A two-consecutive-SLB-days case occurred on Nov. 26–27, with the wind direction transformed evidently between the land breeze and sea breeze. As one class of non-SLB cases, a local pollution case (LP case) that occurred during Dec. 11–12 was chosen in this study. The wind in the LP case came from the sea in the period 13:00–20:00, similar to the SLB case, but the wind in the LP case was quite still in the period 1:00–8:00. On the whole, the average wind speed was 1.8  $\text{m s}^{-1}$  for the LP case relative to 2.8  $\text{m s}^{-1}$  for the SLB case, suggesting that both of the cases were local pollution events. The diurnal variations of meteorological parameters like *T*, RH, BLH, and UV were consistent for the two cases. However, the peak of *T*, BLH, UV, and the valley of RH during the daytime were more marked in the SLB case than the LP case, which is consistent with the discussion in section 3.2.

The LP case here had a higher average NO<sub>2</sub> concentration than the SLB case. As expected, there was an increase in NO<sub>2</sub> concentrations in the evening rush hours for both cases. However, continuously north-easter wind direction (LB) from the midnight to early morning in the SLB case was favor for the diffusion of NO<sub>2</sub>. Whereas, the still wind in the same period of the LP case mostly resulted in accumulation of NO<sub>2</sub> after it emitted from traffic sources. The afternoon peak of O<sub>3</sub> in the LP case

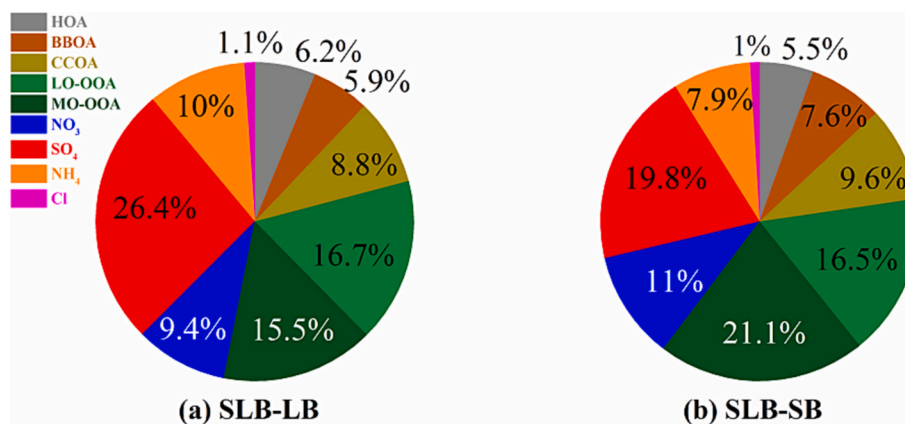


Fig. 5. The fraction of chemical composition in NR-PM<sub>1</sub> during (a) SLB-LB and (b) SLB-SB periods.

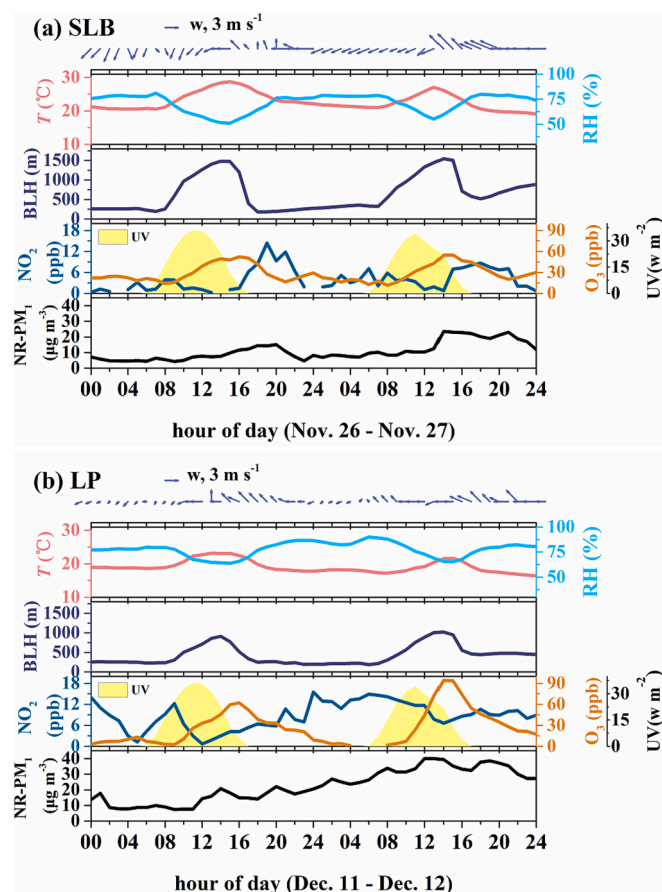


Fig. 6. Temporal variations of meteorological parameters, gaseous pollutants, and NR-PM<sub>1</sub> during (a) the SLB case, and (b) the LP case in 2020.

was more marked than that in the SLB case, especially for the second day of the LP case. The possible reason might be that the high level of NO<sub>2</sub> in the early morning of the LP case provided sufficient precursors for O<sub>3</sub> formation.

The average concentrations of NR-PM<sub>1</sub> were 22.5  $\mu\text{g m}^{-3}$  for the LP case and 10.9  $\mu\text{g m}^{-3}$  for the SLB case. In terms of chemical composition, the NR-PM<sub>1</sub> in the SLB case was characterized by a high fraction of OA (56.7%) and a low fraction of NO<sub>3</sub> (13.4%) compared to those in the LP case (46.7% and 30.6% for OA and NO<sub>3</sub>, respectively). This result is well consistent with the general difference between SLB and non-SLB days discussed above. The whole variation trend of NR-PM<sub>1</sub> concentrations

and its chemical composition was significantly different between the two cases (Fig. 7). For the LP case, the NR-PM<sub>1</sub> concentrations rose continuously (Fig. 7b). The contribution of NO<sub>3</sub> elevated most obviously with the increase of NR-PM<sub>1</sub>. The large fraction of NO<sub>3</sub> reaching 40% appeared from the evening of the first day to the morning of the second day. More abundant precursor NO<sub>2</sub> and the lower  $T$  and higher RH during the night of the LP case enhanced the formation and the partitioning of nitrate from gas to particulate phase, resulting in the continuous increase of NO<sub>3</sub>.

Unlike the LP case, the diurnal variations of NR-PM<sub>1</sub> for the two days of the SLB case were nearly duplicated, with a low level in the early morning, an upward trend during the day, and a peak in the late evening (Fig. 7a). The NR-PM<sub>1</sub> concentration on the second day was nearly twice of that on the first day. Note that the concentrations of chemical composition in the SLB case increased in the comparable pace on the second day compared to the first day. The extremely large fraction of OA in NR-PM<sub>1</sub> in the SLB case was mainly contributed by OOA. As shown in Fig. 7c, the fraction of LO-OOA in NR-PM<sub>1</sub> was relatively stable an entire day, but the fraction of MO-OOA obviously elevated at midday, indicating that the enhanced photochemical formation of MO-OOA during the daytime. The NO<sub>3</sub> fraction in the SLB case increased distinctly in the evening, especially during the second day. It was mainly caused by the synergistic contributions of sufficient precursors emitted by traffic in the evening rush hours, the shallow boundary layer, and likely cyclic amplification, according to the above discussion. Studies on both observations and simulations implied that increased stability caused by aerosol-PBL interaction might continue to affect the atmospheric stratification and aggravate the pollution on the next day (Huang et al., 2018). Additionally, the air mass was blown to the ocean by LB, spent several hours in the marine environment, and then was carried back to the land by SB. Thus, the successive SLB circulations would result in multiple accumulations of pollutants.

#### 4. Conclusion

This study investigated the characteristics of meteorology, air pollutants, and NR-PM<sub>1</sub> chemical composition under the influence of SLB in a coastal city of Southeast China. The diurnal variations of meteorological parameters were similar between SLB and non-SLB days, but the day-night difference in  $T$ , RH, BLH, and UV was larger on SLB days. The diurnal variation of air pollutants, especially NO<sub>2</sub> and NR-PM<sub>1</sub> on SLB days showed an outstanding peak in the late evening, which was likely the combination results of pollutant emissions during the evening rush hours, the sharp reduction of BLH, and the air circulation driven by the daily LB/SB reversals. In addition, the late evening peak of NR-PM<sub>1</sub> was also contributed by secondary formation with large precursor NO<sub>2</sub> emissions.

Compared to non-SLB days, the fraction of OOA in NR-PM<sub>1</sub> increased

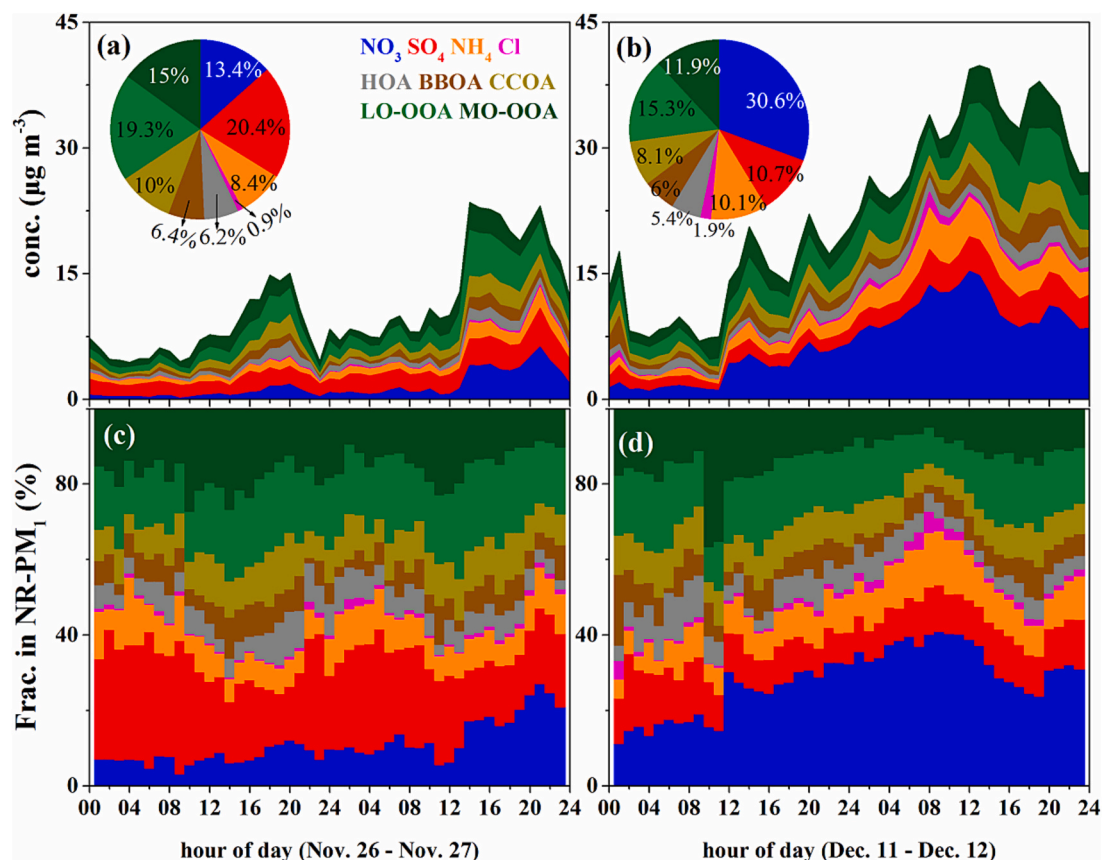


Fig. 7. Temporal variations of mass concentrations and chemical composition of NR-PM<sub>1</sub> during (a,c) the SLB case, and (b,d) the LP case in 2020.

and the fraction of NO<sub>3</sub> decreased significantly on SLB days. SLB days were characterized by strong photochemical indicators such as UV and O<sub>x</sub> during the daytime, which enhanced the photochemical formation of OOA. Moreover, the elevated fraction of sulfate during the LB period was likely due to nocturnal aqueous-phase reaction formation, while the elevated fraction of MO-OOA during the SB period was mainly ascribed to the favorable environment for photochemical processes and aerosol aging. Cases study further highlighted that, compared with LP case, the NR-PM<sub>1</sub> in SLB case was characterized by elevated fraction of OOA, and the successive SLB circulations would result in the cyclic amplification of air pollutants. Our results provide a plausible explanation for the evolution of atmospheric aerosol chemistry and atmospheric pollution processes in coastal cities.

#### CRedit authorship contribution statement

**Yuping Chen:** Conceptualization, Methodology, Software, Investigation, Writing – review & editing. **Chen Yang:** Conceptualization, Writing – review & editing. **Lingling Xu:** Conceptualization, Writing – review & editing. **Xiaolong Fan:** Investigation, Writing – review & editing. **Jiayan Shi:** Visualization. **Ronghua Zheng:** Data curation. **Youwei Hong:** Investigation. **Mengren Li:** Investigation. **Taotao Liu:** Investigation. **Gaojie Chen:** Data curation. **Liqian Yin:** Data curation. **Jinsheng Chen:** Conceptualization, Supervision, Writing – review & editing.

#### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

#### Data availability

Data will be made available on request.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.atmosres.2023.106626>.

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